Praxair Material Safety Data Sheet

1. Chemical Product and Company Identification				
Product Name: Carbon monoxide, compressed (MSDS No. P-4576-J)	Trade Names: Carbon Monoxide			
Chemical Name: Carbon monoxide	Synonyms: Carbonic oxide, carbon oxide			
Chemical Family: Permanent gas	Product Grades: 1.85, 2.5, Ultra High Purity – 3.0, Research - 4.0			
Telephone: Emergencies: 1-800-645-4633* Company Name: Praxair, Inc. CHEMTREC: 1-800-424-9300* 39 Old Ridgebury Road Routine: 1-800-PRAXAIR Danbury, CT 06810-5113 *Call emergency numbers 24 hours a day only for spills, leaks, fire, exposure, or accidents involving this product. For routine information, contact your supplier, Praxair sales representative, or call 1-800-PRAXAIR (1-800-772-9247).				
2. Hazards Identification				

DANGER! Poisonous, flammable, odorless high-pressure gas. Acts on blood, causing damage to central nervous system (CNS). Can be fatal even with adequate oxygen. Can form explosive mixtures with air. Harmful if inhaled. Self-contained breathing apparatus must be worn by rescue workers. Under ambient conditions, this product is a colorless, odorless gas.

OSHA REGULATORY STATUS: This material is considered hazardous by the OSHA Hazard Communications Standard (29 CFR 1910.1200).

POTENTIAL HEALTH EFFECTS:

Effects of a Single (Acute) Overexposure

Inhalation. Depending on the concentration and duration of exposure, may cause headache, drowsiness, dizziness, excitation, rapid breathing, pallor, cyanosis, excess salivation, nausea, vomiting, hallucinations, confusion, angina, convulsions, and unconsciousness. With well-established poisoning, the mucosal surface will be bright red (cherry red). Lack of oxygen can kill.

Skin Contact. No harm expected.

Swallowing. An unlikely route of exposure. This product is a gas at normal temperature and pressure.

Eye Contact. No harm expected.

A vertical line in the left margin indicates revised or new material.

Effects of Repeated (Chronic) Overexposure. Repeated hypoxia from carbon monoxide exposure will cause gradually increasing CNS damage, with loss of sensation in the fingers, poor memory, and mental deterioration. Chronic exposure may facilitate development of atherosclerosis.

Other Effects of Overexposure. Other effects include embryotoxicity, impaired cardiovascular function, pulmonary edema, pneumonia, gross neuropsychiatric damage, memory impairment, permanent CNS damage, and cerebral edema with irreversible brain damage. Late, fatal demyelinization is a rare, but possible, complication.

Medical Conditions Aggravated by Overexposure. Hypoxia from carboxyhemoglobin formation may aggravate established coronary and cerebral circulatory insufficiency.

CARCINOGENICITY: Carbon monoxide is not listed by NTP, OSHA, and IARC.

POTENTIAL ENVIRONMENTAL EFFECTS: No known effects. For further information, see section 12, Ecological Information.

3. Composition/Information on Ingredients

See section 16 for important information about mixtures.

COMPONENT	CAS NUMBER	CONCENTRATION
Carbon Monoxide	630-08-0	>99%*
*The symbol > means "greater than."		

4. First Aid Measures

INHALATION: Immediately remove to fresh air. If not breathing, give artificial respiration with supplemental oxygen given by qualified personnel. If breathing, qualified personnel should give oxygen. Call a physician.

SKIN CONTACT: Wash skin with soap and water. If irritation persists or contact has been prolonged, call a physician.

SWALLOWING: An unlikely route of exposure. This product is a gas at normal temperature and pressure.

EYE CONTACT: An unlikely route of exposure. Flush with water. Hold the eyelids open and away from the eyeballs to ensure that all surfaces are thoroughly flushed. Get medical attention if discomfort persists.

NOTES TO PHYSICIAN: There is no specific antidote. Treatment of overexposure should be directed at the control of symptoms and the clinical condition of the patient. Angina and depression of the ST segment of the electrocardiogram indicate myocardial hypoxia. Exposure to high concentrations can result in cerebral edema. With severe doses, the use of hyperbaric oxygen may be beneficial. Individuals repeatedly overexposed may present positive Romberg's sign.

Contact the Poison Control Center in your area for additional information on patient management and follow-up.

5. Fire Fighting Measures

FLAMMABLE PROPERTIES: Cannot be detected by odor. Forms explosive mixtures with air and oxidizing agents.

SUITABLE EXTINGUISHING MEDIA: CO₂, dry chemical, water spray, or fog.

PRODUCTS OF COMBUSTION: Carbon dioxide

PROTECTION OF FIREFIGHTERS: DANGER! Poisonous, flammable, odorless highpressure gas. Evacuate all personnel from danger area. Do not approach area without selfcontained breathing apparatus. Immediately cool cylinders with water spray from maximum distance, taking care not to extinguish flames. Remove ignition sources if without risk. If flames are accidentally extinguished, explosive reignition may occur. Stop flow of gas if without risk, while continuing cooling water spray. Remove all cylinders from area of fire if without risk. Allow fire to burn out. On-site fire brigades must comply with OSHA 29 CFR 1910.156.

Specific Physical and Chemical Hazards. Heat of fire can build pressure in cylinder and cause it to rupture. No part of cylinder should be subjected to a temperature higher than 125°F (52°C). Carbon monoxide cylinders are equipped with a pressure relief device. (Exceptions may exist where authorized by DOT.) If leaking carbon monoxide catches fire, do not extinguish flames. Flammable and toxic vapors may spread from leak and could explode if reignited by sparks or flames. Explosive atmospheres may linger. Before entering area, especially confined areas, check with an appropriate device.

Protective Equipment and Precautions for Firefighters. Firefighters should wear selfcontained breathing apparatus and full fire-fighting turnout gear.

6. Accidental Release Measures

STEPS TO BE TAKEN IF MATERIAL IS RELEASED OR SPILLED:

DANGER! Poisonous, flammable, odorless high-pressure gas.

Personal Precautions. Cannot be detected by odor. Immediately evacuate all personnel from danger area. Do not approach area without self-contained breathing apparatus. May form explosive mixtures with air. Toxic, flammable gas may spread. Before entering area, especially a confined area, check atmosphere with an appropriate device. Remove all sources of ignition if without risk. Reduce gas with fog or fine water spray. Shut off flow if without risk. Ventilate area or move cylinder to well-ventilated area.

Environmental Precautions. Prevent waste from contaminating the surrounding environment. Keep personnel away. Discard any product, residue, disposable container, or liner in an environmentally acceptable manner, in full compliance with federal, state, and local regulations. If necessary, call your local supplier for assistance.

7. Handling and Storage

PRECAUTIONS TO BE TAKEN IN HANDLING: May form explosive mixtures with air. Protect cylinders from damage. Use a suitable hand truck to move cylinders; do not drag, roll, slide, or drop. Keep away from heat, sparks, and open flame. Use only spark-proof tools and explosion-proof equipment. Never attempt to lift a cylinder by its cap; the cap is intended solely to protect the valve. Never insert an object (e.g., wrench, screwdriver, pry bar) into cap openings; doing so may damage the valve and cause a leak. Use an adjustable strap wrench to remove over-tight or rusted caps. Open valve slowly. If valve is hard to open, discontinue use and contact

your supplier. Close valve after each use; keep closed even when empty. For other precautions in using carbon monoxide, see section 16.

PRECAUTIONS TO BE TAKEN IN STORAGE: Store and use with adequate ventilation. Protect cylinders from direct sunlight. Separate carbon monoxide cylinders from oxygen and other oxidizers by at least 20 ft (6.1 m), or use a barricade of noncombustible material. This barricade should be at least 5 ft (1.53 m) high and have a fire resistance rating of at least ½ hr. Firmly secure cylinders upright to keep them from falling or being knocked over. Screw valve protection cap firmly in place by hand. Store only where temperature will not exceed 125°F (52°C). Store full and empty cylinders separately. Use a first-in, first-out inventory system to prevent storing full cylinders for long periods. Post "No Smoking or Open Flames" signs in storage and use areas. There must be no sources of ignition. All electrical equipment in storage areas must be explosion-proof. Storage areas must meet national electric codes for Class 1 hazardous areas.

RECOMMENDED PUBLICATIONS: For further information on storage, handling, and use of this product, see Praxair publications P-14-153, *Guidelines for Handling Gas Cylinders and Containers,* and P-15-437, *Safe Handling of Carbon Monoxide*. Obtain from your local supplier.

8. Exposure Controls/Personal Protection

COMPONENT	OSHA PEL	ACGIH TLV-TWA (2009)
Carbon Monoxide	50 ppm	25 ppm

TLV-TWAs should be used as a guide in the control of health hazards and not as fine lines between safe and dangerous concentrations.

IDLH = 1200 ppm

ENGINEERING CONTROLS:

Local Exhaust. Use an explosion-proof local exhaust system with sufficient air flow to keep the carbon monoxide concentration below the applicable exposure limits in the worker's breathing zone.

Mechanical (General). Not recommended as a primary ventilation system to control worker's exposure.

Special. None

Other. None

PERSONAL PROTECTIVE EQUIPMENT:

Skin Protection. Wear work gloves when handling cylinders.

Eye/Face Protection. Wear safety glasses when handling cylinders. Select per OSHA 29 CFR 1910.133.

Respiratory Protection. A respiratory protection program that meet OSHA 29 CFR 1910.134, ANSI Z88.2, or MSHA 30 CFR 72.710 (where applicable) requirements must be followed whenever workplace conditions warrant respirator use. Use an air-supplied or air-purifying cartridge if the action level is exceeded. Ensure the respirator has the appropriate protection factor for the exposure level. If cartridge type respirators are used, the cartridge must be appropriate for the chemical exposure (e.g., an organic vapor cartridge). For emergencies or instances with unknown exposure levels, use a self-contained breathing apparatus.

9. Physical and Chemical Properties				
APPEARANCE:	Colorless gas			
ODOR:	Odorless			
ODOR THRESHOLD:	Not applicable.			
PHYSICAL STATE:	Gas at normal temperature and pressure			
pH:	Not applicable.			
MELTING POINT at 1 atm:	-337°F (-205.0°C)			
BOILING POINT at 1 atm:	-312.61°F (-191.45°C)			
FLASH POINT (test method):	Flammable gas, not applicable.			
EVAPORATION RATE (Butyl Acetate = 1):	Not applicable.			
FLAMMABILITY:	Flammable			
FLAMMABLE LIMITS IN AIR, % by volume:	LOWER: 12.5% UPPER: 74%			
VAPOR PRESSURE:	Not applicable.			
VAPOR DENSITY at 68°F (20°C) and 1 atm:	0.0725lb/ft ³ (1.161kg/m ³)			
SPECIFIC GRAVITY (H ₂ O = 1):	Not applicable.			
SPECIFIC GRAVITY (Air = 1) at 70°F (21.1°C)				
and 1 atm:	0.9676			
SOLUBILITY IN WATER , vol/vol at 32°F (0°C)	0.035			
and 1 atm:				
PARTITION COEFFICIENT: n-octanol/water:	Not available.			
AUTOIGNITION TEMPERATURE:	1128°F (608.9°C)			
DECOMPOSITION TEMPERATURE:	752°F (400°C)			
PERCENT VOLATILES BY VOLUME:	100			
MOLECULAR WEIGHT:	28.01			
MOLECULAR FORMULA:	CO			

10. Stability and Reactivity

CHEMICAL STABILITY:
Unstable
Stable

CONDITIONS TO AVOID: Temperatures above 752°F (400°C)

INCOMPATIBLE MATERIALS: Oxidizing agents, oxygen, flammables, metal oxides, halogenated fluorides, metals in the presence of moisture and/or sulfur compounds

HAZARDOUS DECOMPOSITION PRODUCTS: Carbon monoxide will decompose above 752°F (400°C) to form carbon dioxide and carbon.

POSSIBILITY OF HAZARDOUS REACTIONS: 🖂 May Occur 🗌 Will Not Occur

Forms explosive mixtures with air and oxidizing agents.

11. Toxicological Information

ACUTE DOSE EFFECTS: $LC_{50} = 3760$ ppm, 1 hr rat.

STUDY RESULTS: Carbon monoxide produces embryofetal toxicity in laboratory animals but only at doses that cause maternal toxicity. There is no information available on possible effects in humans.

12. Ecological Information

ECOTOXICITY: No information available on ecological effects.

OTHER ADVERSE EFFECTS: Carbon monoxide does not contain any Class I or Class II ozone-depleting chemicals.

13. Disposal Considerations

WASTE DISPOSAL METHOD: Do not attempt to dispose of residual or unused quantities. Return cylinder to supplier.

14. Transport Information

DOT/IMO S	SHIP	PING NAME:	Carbon m	onoxide, compress	ed		
HAZARD		PACKING		IDENTIFICATION		PRODUCT	
CLASS:	2.3	GROUP/Zone:	NA/D	NUMBER:	UN1016	RQ:	None
SHIPPING LABEL(s): POISON GAS, FLAMMABLE GAS*							
PLACARD	(wh	en required):	POISON (GAS, FLAMMABLE	GAS*		

*The words in the POISON GAS diamond are INHALATION HAZARD.

SPECIAL SHIPPING INFORMATION: Cylinders should be transported in a secure position, in a well-ventilated vehicle. Cylinders transported in an enclosed, non-ventilated compartment of a vehicle can present serious safety hazards.

Additional Marking Requirement: INHALATION HAZARD

Shipment of compressed gas cylinders that have been filled without the owner's consent is a violation of federal law [49 CFR 173.301(b)].

MARINE POLLUTANTS: Carbon monoxide is not listed as a marine pollutant by DOT.

15. Regulatory Information

The following selected regulatory requirements may apply to this product. Not all such requirements are identified. Users of this product are solely responsible for compliance with all applicable federal, state, and local regulations.

U.S. FEDERAL REGULATIONS:

EPA (ENVIRONMENTAL PROTECTION AGENCY)

CERCLA: COMPREHENSIVE ENVIRONMENTAL RESPONSE, COMPENSATION, AND LIABILITY ACT OF 1980 (40 CFR Parts 117 and 302):

Reportable Quantity (RQ): None

SARA: SUPERFUND AMENDMENT AND REAUTHORIZATION ACT:

SECTIONS 302/304: Require emergency planning based on Threshold Planning Quantity (TPQ) and release reporting based on Reportable Quantities (RQ) of Extremely Hazardous Substances (EHS) (40 CFR Part 355):

TPQ: None EHS RQ (40 CFR 355): None **SECTIONS 311/312:** Require submission of MSDSs and reporting of chemical inventories with identification of EPA hazard categories. The hazard categories for this product are as follows:

IMMEDIATE: Yes DELAYED: Yes PRESSURE: Yes REACTIVITY: No FIRE: Yes

SECTION 313: Requires submission of annual reports of release of toxic chemicals that appear in 40 CFR Part 372.

Carbon monoxide is not subject to reporting under Section 313.

40 CFR 68: RISK MANAGEMENT PROGRAM FOR CHEMICAL ACCIDENTAL RELEASE PREVENTION: Requires development and implementation of risk management programs at facilities that manufacture, use, store, or otherwise handle regulated substances in quantities that exceed specified thresholds.

Carbon monoxide is not listed as a regulated substance.

TSCA: TOXIC SUBSTANCES CONTROL ACT: Carbon monoxide is listed on the TSCA inventory.

OSHA: OCCUPATIONAL SAFETY AND HEALTH ADMINISTRATION:

29 CFR 1910.119: PROCESS SAFETY MANAGEMENT OF HIGHLY HAZARDOUS CHEMICALS: Requires facilities to develop a process safety management program based on Threshold Quantities (TQ) of highly hazardous chemicals.

Carbon monoxide is not listed in Appendix A as a highly hazardous chemical. However, any process that involves a flammable gas on site in one location in quantities of 10,000 lb (4536 kg) or greater is covered under this regulation unless the gas is used as a fuel.

STATE REGULATIONS:

CALIFORNIA: Carbon monoxide is listed by California under the SAFE DRINKING WATER AND TOXIC ENFORCEMENT ACT OF 1986 (Proposition 65).

WARNING: Carbon monoxide is a chemical known to the State of California to cause birth defects or other reproductive harm. *(California Health and Safety Code §25249.5 <u>et seq.)*</u>

PENNSYLVANIA: Carbon monoxide is subject to the PENNSYLVANIA WORKER AND COMMUNITY RIGHT-TO-KNOW ACT (35 P.S. Sections 7301-7320).

16. Other Information

Be sure to read and understand all labels and instructions supplied with all containers of this product.

OTHER HAZARDOUS CONDITIONS OF HANDLING, STORAGE, AND USE: *Poisonous, flammable, odorless high-pressure gas.* Cannot be detected by odor. Use piping and equipment adequately designed to withstand pressures to be encountered. Use only in a closed system. Use a backflow prevention device in any piping. Ground all equipment. Electrical equipment must be non-sparking or explosion-proof. Never work on a pressurized system. If a leak occurs, close the cylinder valve, blow down the system by venting vapor to a safe place in an environmentally safe manner in compliance with all federal, state, and local laws; then repair the leak. Follow safe practices when returning cylinder to supplier. Be sure

valve is closed; then install valve outlet cap or plug, leak-tight. Never place a compressed gas cylinder where it may become part of an electrical circuit.

NOTE: Prior to using any plastics, confirm their compatibility with carbon monoxide. Avoid using pure nickel. Corrosion of pure nickel in carbon monoxide atmospheres exceeds 50 mil/yr (1.27 mm/yr) at room temperatures.

Mixtures. When you mix two or more gases or liquefied gases, you can create additional, unexpected hazards. Obtain and evaluate the safety information for each component before you produce the mixture. Consult an industrial hygienist or other trained person when you evaluate the end product. Remember, gases and liquids have properties that can cause serious injury or death.

HAZARD RATING SYSTEMS:

NFPA RATINGS	51
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HMIS RATINGS:

HEALTH	= 3	HEALTH	= 1
FLAMMABILITY	= 4	FLAMMABILITY	= 4
INSTABILITY	= 0	PHYSICAL HAZARD	= 3
SPECIAL	= None		

*An asterisk used in conjunction with HMIS health hazard ratings designates a carcinogenic or reproductive hazard.

STANDARD VALVE CONNECTIONS FOR U.S. AND CANADA: THREADED: CGA-350 connection is standard PIN-INDEXED YOKE: NA ULTRA-HIGH-INTEGRITY CONNECTION: CGA-724 NOTE: Do not use a nickel gasket Use the proper CGA connections. DO NOT USE ADAPTERS. Additional limited-standard connections may apply. See CGA pamphlet V-1 listed below.

Ask your supplier about free Praxair safety literature as referred to in this MSDS and on the label for this product. Further information can be found in the following pamphlets published by the Compressed Gas Association, Inc. (CGA), 4221 Walney Road, 5th Floor, Chantilly, VA 20151-2923, Telephone (703) 788-2700, http://www.cganet.com/Publication.asp.

- AV-1 Safe Handling and Storage of Compressed Gases
- P-1 Safe Handling of Compressed Gases in Containers
- V-1 Compressed Gas Cylinder Valve Inlet and Outlet Connections
- Handbook of Compressed Gases, Fourth Edition

Praxair asks users of this product to study this MSDS and become aware of product hazards and safety information. To promote safe use of this product, a user should (1) notify employees, agents, and contractors of the information in this MSDS and of any other known product hazards and safety information, (2) furnish this information to each purchaser of the product, and (3) ask each purchaser to notify its employees and customers of the product hazards and safety information.

The opinions expressed herein are those of qualified experts within Praxair, Inc. We believe that the information contained herein is current as of the date of this Material Safety Data Sheet. Since the use of this information and the conditions of use of the product are not within the control of Praxair, Inc., it is the user's obligation to determine the conditions of safe use of the product.

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